

Subscriber access provided by ISTANBUL TEKNIK UNIV

Three New Heptaprenylhydroguinone Derivatives from the Sponge Ircinia fasciculata

Y. Venkateswarlu, and M. Venkata Rami Reddy

J. Nat. Prod., 1994, 57 (9), 1286-1289• DOI: 10.1021/np50111a018 • Publication Date (Web): 01 July 2004

Downloaded from http://pubs.acs.org on April 4, 2009

More About This Article

The permalink http://dx.doi.org/10.1021/np50111a018 provides access to:

- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article



Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

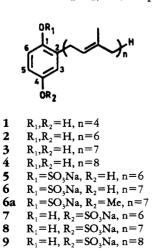
THREE NEW HEPTAPRENYLHYDROQUINONE DERIVATIVES FROM THE SPONGE IRCINIA FASCICULATA¹

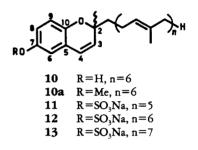
Y. VENKATESWARLU* and M. VENKATA RAMI REDDY

Organic Chemistry Division-I, Indian Institute of Chemical Technology, Hyderabad 500007, India

ABSTRACT.—2-Heptaprenylhydroquinone [3], its sulfate [6], 2-(hexaprenylmethyl)-2methylchromenol [10], and 2-(hexaprenylmethyl)-2-methylchromanol [14] have been isolated from the sponge *Ircinia fasciculata*. Three of these compounds [6, 10, 14] are novel. Their structures have been established by spectral methods and chemical transformations.

In the course of our search for bioactive metabolites from Indian marine invertebrates (1-3), we investigated the sponge Ircinia fasciculata Pallas (Spongillidae) collected on the coast of the Gulf of Mannar, India. In previous reports the genus Ircinia has been shown to produce furanosesterterpenes (4-6) and polyprenvlated hydroquinones (7-8) [1-4]. Recently, sulfated derivatives of polyprenvlated hydroquinones [5, 7-9] and the chromenols 11-13 have been isolated from a Dysidea species (9) and Sarcotragus spinulosus (10). These sulfated metabolites have been shown to inhibit H^+, K^+ and Na^+, K^+ -ATPase. Chromatographic separation of the crude extract of I. fasciculata afforded the three new compounds 2-heptaprenylhydroquinone-1sulfate [6]. 2-(hexaprenvlmethyl)-2methylchromenol [10], 2-(hexaprenyl-



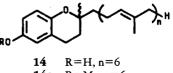


methyl)-2-methylchromanol [14], and the known heptaprenylhydroquinone, 3.

The MeOH-CHCl₂ (1:1) extract of the lyophilized sponge *Ircinia fasciculata* was partitioned between EtOAc and H₂O. The EtOAc extract was subjected to Si gel chromatography eluting with hexane, hexane/EtOAc mixtures, and C₆H₆/ Me₂CO mixtures, and afforded three less polar compounds, **3**, **10**, **14**, and one highly polar compound, **6**.

Compound **3** was obtained as an oil, eims $[M]^+$ m/z 586, and analyzed for $C_{41}H_{62}O_3$. The study of its uv, ir, ¹H- and ¹³C-nmr spectral data (7,11) led to its identification as 2-heptaprenyl-1,4-hydroquinone, which has previously been isolated from *Ircinia spinosula* (7) and *Hippospongia communis* (11).

Compound 10 was obtained as a viscous oil, $[\alpha]D + 3.6^{\circ}$ (c=1, CHCl₃), eims $[M]^{+}$ m/z 584, and analyzed for



14a R=Me, n=6 **15** R=H, n=1-6

 $C_{41}H_{60}O_2$ by microanalysis. The peaks at 3420, 1580, and 1485 cm⁻¹ in its ir spectrum indicated the presence of hydroxyl groups and an aromatic moiety in the molecule, which was further confirmed by formation of a monomethyl ether [10a] (CH₃I/K₂CO₃). Its uv spectrum is typical of a chromene (12), and this structure was supported by ¹³C-nmr signals at δ 149.3 s (C-10), 146.6 s (C-7), 130.6 d (C-3), 122.6 d (C-4), 121.8 s (C-5), 116.5 d (C-6), 115.4 d (C-9), 112.9 d (C-8), and 78.0 s (C-2), as well as a base peak in its mass spectrum at m/z 161 corresponding to the 2-methyl-7hydroxychromene ion. In the ¹H-nmr spectrum of compound **10**, the signals at δ 6.24 (d, J=10 Hz) and 5.57 (d, J=10 Hz) were assigned to olefinic protons of the chromene system. The signals at δ 6.61 (d, J=9 Hz), 6.53 (dd, J=9 and 3 Hz), and 6.42 (d, J=3 Hz) were due to aromatic protons. Further, the ¹H-nmr spectrum of 10 showed signals integrating for six olefinic protons at δ 5.10 and eight methyls at δ 1.66 (s, CH₃), 1.58 (br s, $6 \times CH_3$), and 1.35 (s, CH_3). From the foregoing spectral data, the structure of compound 10 was established as 2-(hexaprenylmethyl)-2-methylchromenol [10]. An all-trans polyprenylic side-chain was recognized by comparison with previously reported data (7). Recently, the sulfate derivative [12] of compound 10 was isolated from Sarcotragus spinulosus (10).

Compound 14 was obtained as an oil, $[\alpha]D + 8^{\circ}$ (c=0.11, CHCl₃), eims $[M]^+ m/z 586$, and analyzed for $C_{41}H_{62}O_2$ by microanalysis. Its ir spectrum showed peaks at 3410, 1665, 1620, and 1500 cm⁻¹ consistent with the presence of phenolic hydroxyl and aromatic moieties. On methylation (CH₃I/K₂CO₃), compound 14 formed a monomethyl ether [14a], thus confirming the presence of a phenolic hydroxyl group. The ¹H-nmr spectrum of compound 14 was almost identical with that of 10, except for the absence of the olefinic protons of the chromene moiety. It was evident from

the mass spectrum that compound 14 is a dihydro derivative of 10, which was supported by a fragment ion in the mass spectrum at m/z 163, corresponding to the 7-hydroxy-2-methylbenzopyran ion. Compound 14 was previously isolated as part of a mixture of chromanols [15] from the digestive glands of a dorid nudibranch, *Dendrodoris grandiflora* (13), but was not fully characterized.

Compound 6 was obtained as a colorless gum, positive fabms [M+Na]⁺ m/z 711, and analyzed for C₄₁H₆₁O₅SNa by microanalysis. The uv, ir, and ¹H- and ¹³C-nmr spectra were almost similar to those of compound 5, which was isolated from Dysidea sp. (9). However, compound 6 was a higher homologue of one isoprene unit and found to be a new compound. Further, it was observed that compound **6**, on standing for long periods in $CHCl_2$ or when in contact with Si gel, converted into compounds 3 and 14 in a 2:1 ratio. suggesting that compounds 3 and 14may be artifacts obtained during the isolation process.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—¹H-Nmr (200 MHz) and ¹³C-nmr (50 MHz) spectra were recorded on a Varian Gemini spectrometer using TMS as internal standard. Chemical shifts are reported in ppm values and coupling constants (J) are expressed in Hz. Elemental analysis was carried out on a Perkin-Elmer 240C instrument. Optical rotations were measured with a Jasco Dip-370 polarimeter. Uv and ir spectra were recorded on Shimadzu and Perkin-Elmer 1310 spectrophotometers, respectively. Eims and fabms were recorded on Finnigan MAT 1020 and JEOLSX 102/ DA-6000 instruments, respectively.

COLLECTION, EXTRACTION AND ISOLATION.— The sponge *Ircinia fasciculata* was collected in April 1993 at a depth of 40 feet by scuba diving, near Tuticorin in the Gulf of Mannar, India. A voucher specimen (IIC-124) has been lodged at the museum of the National Institute of Oceanography, Goa, India. The specimens were cut into thin slices and soaked in MeOH at the site of collection until work up. After the removal of MeOH, the sponge (500 g dry wt) was lyophilized and extracted with MeOH-CH₂Cl₂(1:1) (2 liters). After evaporation of solvent under reduced pressure, the crude extract was partitioned between EtOAc and H₂O. Concentration of the organic layer resulted in a gummy crude extract (16 g), which was subjected to Si gel (100–200 mesh) cc, eluting with hexane, hexane/EtOAc mixtures and $C_6H_6/$ Me₂CO mixtures, to give four compounds, **3** (60 mg), **6** (400 mg), **10**, (500 mg), and **14** (28 mg).

2-HEPTAPRENYLHYDROQUINONE-1-SULFATE [6].—Compound 6 has obtained as a colorless gum; anal., found C 71.49%, H 8.98%, S 4.70%, required for C41H61O5SNa, C 71.47%, H 8.92%, S 4.65%; uv λ max (CHCl₃) nm (ε) 275 (3200), 241 (4900) nm; ir v max (neat) 3440, 2910, 1660, 1610, 1580, 1480, 1440, 1255, 1040, 850, 735 cm^{-1} ; ¹H nmr (CDCl₃) δ 7.11 (1H, d, J=9 Hz), 6.60 (1H, br s), 6.52 (1H, br d, J=9 Hz), 5.20 (1H, brs), 5.10(6H, m), 3.36(2H, brs), 1.9-2.15 (24H, m, 12×CH₂), 1.66 (3H, s, CH₃), 1.58 (21H, br s, 7×CH₃); ¹³C nmr (CDCl₃) δ 153.9 s, 147.7 s, 137.4 s, 136.3 s, 135.2 s, 134.8 s (4C), 131.1 s, 124.0 d (6C), 123.0 d, 121.3 d, 116.5 d, 113.4 d, 39.7 t (7C), 28.2 t, 26.5 t (5C), 25.8 q, 17.8 q, 16.0 q (6C); fabms m/z 711 [(M+Na)⁺].

Methylation of compound **6**.—Compound **6** (10 mg) was refluxed in dry Me₂CO (8 ml) with K₂CO₃ (300 mg) and methyl iodide (1 ml) for 12 h. After the usual workup, the product **6a** (8 mg) was purified by Si gel cc. Anal., found C 71.81%, H 9.20%, S 4.57%, required for C₄₂H₆₃O₃SNa, C 71.76%, H 9.03%, S 4.56%; uv λ max (CHCl₃)(ϵ) 276 (3200), 241 (4800) nm; ir ν max (neat) 2910, 1650, 1600, 1470, 1440, 1250, 1040, 850, 730 cm⁻¹; ¹H nmr (CDCl₃) δ 7.28 (1H, d, J=9 Hz), 6.63 (1H, br d, J=9 Hz), 6.62 (1H, br s), 5.25 (1H, br t, J=7 Hz), 5.1 (6H, m), 3.61 (3H, s, OCH₃), 3.38 (2H, br d, J=7 Hz), 1.9–2.15 (24H, m, 12×CH₂), 1.66 (3H, s, CH₃), 1.58 (21H, br s, 7×CH₄); fabms m/z 725 [(M+Na)⁺].

2-(HEXAPRENYLMETHYL)-2-METHYL-CHROMENOL [10].-Compound 10 was obtained as a colorless oil, $[\alpha]D + 3.6^{\circ} (c=1, CHCl_3); anal.,$ found C 84.24%, H 10.48%, required for C₄₁H₆₀O₂, C 84.19%, H 10.34%; uv λ max (CHCl₃) (ε) 330 (3800), 264 (5100), 243 (13,800) nm; ir v max (neat) 3420, 2920, 1580, 1485, 1450, 1220 cm⁻¹; ¹H nmr (CDCl₃) δ 6.61 (1H, d, J=9 Hz), 6.53 (1H, dd, J=9 and 3 Hz), 6.42 (1H, d, J=3 Hz), 6.24 (1H, d, J=10 Hz), 5.57 (1H, d, J=10 Hz),5.1 (6H, m), 1.9-2.15 (24H, m, 12×CH₂), 1.66 $(3H, s, CH_3), 1.58 (18H, br s, 6 \times CH_3), 1.35 (3H,$ s, CH₃); ¹³C nmr (CDCl₃) δ 149.3 s, 146.6 s, 135.1 s, 134.7 s (5C), 131.0 s, 130.6 d, 124.2 d (5C), 123.9 d, 122.6 d, 121.8 s, 116.7 d, 115.4 d, 112.9 d, 78.6 s, 40.7 t, 39.6 t (4C), 26.5 t (5C), 25.8 q, 25.6 q, 22.5 t, 17.5 q, 15.9 q (5C); eims m/z 584 (M⁺), 515, 447, 379, 311, 243, 175, 161 (base peak), 69.

Methylation of compound **10**.—Compound **10** (50 mg) was refluxed in dry Me₂CO (10 ml) with K_2CO_3 (1 g) and methyl iodide (2 ml) for 12 h. After the usual workup, the product **10a** (50 mg) was purified by Si gel cc. Anal., found C 84.31%, H 10.49%, required for $C_{42}H_{62}O_2$, C 84.22%, H 10.43%; $[\alpha]D + 0.4^{\circ}$ (c=1, CHCl₃); uv λ max (CHCl₃)(ε) 331 (3500), 263 (5400), 244 (11,800) nm; ir ν max (neat) 2920, 1580, 1490, 1450, 1260, 1040 cm⁻¹; ¹H nmr (CDCl₃) δ 6.68 (1H, d, J=9 Hz), 6.63 (1H, dd, J=9 and 3 Hz), 6.51 (1H, d, J=3 Hz), 6.30 (1H, d, J=10 Hz), 5.58 (1H, d, J=10 Hz), 5.10 (6H, m), 3.72 (3H, s, OCH₃), 1.9–2.15 (24H, m, 12×CH₂), 1.66 (3H, s, CH₃), 1.58 (18H, m, 6×CH₃), 1.37 (3H, s, CH₃); eims m/z 598 (M⁺), 175 (base peak), 69.

2-(HEXAPRENYLMETHYL)-2-METHYL-CHROMANOL [14].—Compound 14 was obtained as an oil, $[\alpha]D + 8^{\circ} (c=0.11, CHCl_3)$; anal., found C 84.02%, H 10.74%, required for C41H62O2, C 83.9%, H 10.65%; uv λ max (CHCl₃) (ε) 298 (4300), 242 (8500) nm; ir v max (neat) 3410, 2920, 1665, 1620, 1500, 1450, 1380, 1220, 820 cm^{-1} ; ¹H nmr (CDCl₃) δ 6.62 (1H, d, J=9 Hz), 6.55 (1H, dd, J=9 and 3 Hz), 6.51 (1H, d, J=3 Hz), 5.1 (6H, m), 2.71 (2H, t, J=7.5 Hz), 1.9- $2.15(24H, m, 12 \times CH_2), 1.78(2H, t, J=7.5 Hz),$ 1.66 (3H, s, CH₃), 1.58 (18H, br s, 6×CH₃), 1.27 $(3H, s, CH_3)$; ¹³C nmr (CDCl₃) δ 148.5, 148.1, 135.3, 134.9 (4C), 131.4, 124.3 (5C), 122.0, 117.9, 115.4, 114.5, 75.6, 39.7 (4C), 39.4, 31.0, 26.7 (5C), 25.7, 24.2, 22.4, 22.2, 17.9, 15.0 (5C); eims m/z 586 (M⁺), 517, 449, 381, 313, 245, 177, 163, 69 (base peak).

Methylation of compound 14.—Compound 14 (8 mg) was refluxed in dry Me₂CO (8 ml) with K₂CO₃ (300 mg) and methyl iodide (1 ml) for 12 h. After the usual workup, the product 14a (8 mg) was purified by Si gel cc. Anal., found C 84.02%, H 10.78%, required for C42H64O2, C 83.94%, H 10.73%; $[\alpha]D + 0.6^{\circ}$ (c=0.4, CHCl₃); uv λ max (CHCl₃)(e) 297 (4,100), 241 (5,450) nm; ir v max (neat) 2920, 1495, 1450, 1380, 1220, 1040 cm⁻¹ ¹H nmr (CDCl₃) δ 6.72 (1H, d, J=9 Hz), 6.66 (1H, d, J=9 and 3 Hz), 6.60 (1H, d, J=3 Hz),5.10 (6H, m), 3.74 (3H, s, OCH₃), 2.73 (2H, t, J=7.5 Hz), 1.9–2.15 (24H, m, 12 × CH₂), 1.8 $(2H, t, J=7.5 Hz), 1.66 (3H, s, CH_3), 1.58 (18H,$ br s, $6 \times CH_3$), 1.28 (3H, s, CH_3); eims m/z 600 (M⁺), 531, 463, 395, 327, 259, 191, 177, 69 (base peak).

ACKNOWLEDGMENTS

The authors are thankful to Dr. P.A. Thomas for identifying the sponge, the Department of Ocean Development for financial assistance, the Director, IICT, and Dr. J.S. Yadav, Head, Organic Chemistry Division-I, for their encouragement. Thanks are due to RSIC, Lucknow for providing fabms.

LITERATURE CITED

1. M. Venkata Rami Reddy, S. Lakshman,

A.V. Rama Rao, Y. Venkateswariu, and J. Venkateswara Rao, *J. Nat. Prod.*, **56**, 970 (1993).

- Y. Venkateswarlu, M. Venkata Rami Reddy, K.V.N.S. Srinivas, and J. Venkateswara Rao, Ind. J. Chem., **32B**, 704 (1993).
- M. Venkata Rami Reddy, Y. Venkateswarlu, and J. Venkateswara Rao, *Ind. J. Chem.*, 32B, 1196 (1993).
- 4. D.J. Faulkner, *Tetrahedron Lett.*, 3821 (1973).
- F. Cafieri, E. Fattorusso, C. Santacroce, and L. Minale, *Tetrabedron*, 28, 1579 (1972).
- 6. I. Rothberg and P. Shubiak, Tetrabedron Lett., 769 (1975).
- G. Cimino, S. De Stefano, and L. Minale, *Tetrahedron*, 28, 1315 (1972).
- 8. G. Cimino, S. De Stefano, and L. Minale,

Experientia, 28, 1401 (1972).

- N. Fusetani, M. Sugano, S. Matsunaga, K. Hashimoto, H. Shikama, A. Ohta, and H. Nagano, *Experientia*, 43, 1233 (1987).
- V.A. Stonik, T.N. Makarieva, and A.S. Dmitrenok, J. Nat. Prod., 55, 1256 (1992).
- Y.F. Pouchues, J.F. Verbist, J.F. Biard, and K. Boukef, J. Nat. Prod., 51, 188 (1988).
- Y. Venkateswarlu, D.J. Faulkner, J.L. Rios Steiner, E. Corcoran, and J. Clardy, *J. Org. Chem.*, 56, 6271 (1991).
- G. Cimino, S. De Rosa, S. De Stefano, R Morrone, and G. Sodano, *Tetrahedron*, 41, 1093 (1985).

Received 19 January 1994